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[5,5] Sigmatropic shift of N-phenyl-N'-(2-thiazolyl)hydrazines and N,N'-bis(2-thiazolyl)hydrazines into 2-amino-5-(p-aminophenyl)thiazoles and 5,5'-bis(2-aminothiazole) derivatives †

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Abstract

[5,5] Sigmatropic shift of N-phenyl-N'-(2-thiazolyl)hydrazines and N,N'-bis(2-thiazolyl)hydrazines in acid-catalyzed benzidine-type rearrangement into 2-amino-5-(p-aminophenyl)thiazoles and 5,5'-bis(2-aminothiazole) derivatives is described, respectively. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: thiazole; hydrazine; thiourea; benzidine-type rearrangement; acid catalyst.

The acid-catalyzed [5,5] sigmatropic shift of hydrazobenzene to 4,4'-diaminobiphenyl is well known and the related studies were published numerously. We wish to report examples where N-phenyl-N'-(2-thiazolyl)hydrazines and N,N'-bis(2-thiazolyl)hydrazines could perform [5,5] sigmatropic shift in acid-catalyzed benzidine-type rearrangement, respectively.

Because of our interest in the thiazole compounds, we attempted the condensation of N-anilinothiourea $(1)^{2a}$ and α -chloroacetone to give N-phenyl-N'-[2-(4-methyl)thiazolyl]hydrazine (3a) according to the Hantzsch method.³ It was found that when the mixture was refluxed in MeOH under neutral conditions, 2-amino-5-(p-aminophenyl)-4-methylthiazole (5a) was obtained in 81% isolated yield. Otherwise, when it was carried out at room temperature, hydrazine derivative 3a was obtained in 82% isolated yield.

In order to investigate this phenomena stepwise, as illustrated in Scheme 1, a series of hydrazine derivatives (3) have been prepared in high yield by the treatment of 1 with α -chlorocarbonyl compounds for 12–15 h at room temperature or 45°C [pathway (A)].

In the next step of pathway (A) in Scheme 1, compound 3 (5 mmol) was refluxed in 0.5 ml of 35% HCl and 20 ml of H_2O solution to rearrange to compound 5 which, otherwise, was prepared by refluxing directly [pathway (B)] compound 1 and α -chlorocarbonyl compounds in MeOH (refluxed in H^+/H_2O for 5b). The reactions in Scheme 1 are summarized in Table 1.

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[†] Dedicated to Professor T. H. Chan on the occasion of his 60th birthday.

Scheme 1. $Table \ 1$ The reactions of N-anilinothiourea (1) with α -chlorocarbonyl compounds

chlorocarbonyl	reaction condition	product	yield, %a	1 H-nmr data (DMSO- d_{6}), δ
0	MeOH/r.t/12 h	<u>3a</u>	82	2.18 (s, 3H), 6.23 (s, 1H), 6.7-7.2 (m, 5H), 8.15 (s, 1H), 9.17 (s, 1H)
Me CI	MeOH/reflux/12 h	5a	81(78) ^b	(CDCl ₃); 2.2 (s, 3H), 4.2 (bs, 2H), 6.2 (bs, 2H), 6.62 (d, <i>J</i> = 9Hz, 2H), 7.05 (d, <i>J</i> = 9 Hz, 2H)
OMe 	H ⁺ /H ₂ O/45 °C/15 h	<u>3</u> b	78	(CDCl ₃); 7.4 (d, $J = 3.8$ Hz, 1H), 8.05 (d, $J = 3.8$ Hz, 1H), $7.45-8.1$ (m, 7H)
MeO	H ⁺ /H ₂ O/110 ℃/12 h	5b	83(72) ^b	4.0 (bs, 2H), 6.6 (d, <i>J</i> = 9Hz, 2H), 6.7 (bs, 2H), 6.9 (s, 1H), 7.15 (d, <i>J</i> = 9Hz, 2H)
O	MeOH/r.t/12 h	3c	85	6.1 (s, 1H), 6.9-7.75 (m, 12H)
Ph	MeOH/reflux/13 h	5c	88(88) ^b	4.7 (bs, 2H), 6.65 (d, <i>J</i> = 9Hz, 2H), 6.7 (bs, 2H), 6.9 (d, <i>J</i> = 9Hz, 2H), 7.2-7.4 (m, 5H)

 a yield isolated after column chromatography. b yield rearranged to $\underline{\mathbf{5}}$ by refluxing $\underline{\mathbf{3}}$ (H $^{+}$ /H $_{2}$ O).

The structure of **5** was evident from 1 H NMR spectroscopic data. After the rearrangement of **3** to **5**, the thiazole 5-H signal in the region δ 6.1–7.4 of **3** disappeared and the pattern of *N*-phenyl proton peaks changed to doublet and doublet (J=9.0 Hz) in the region δ 6.60–7.15.

From the fact that 5a and 5c could be directly obtained by refluxing 1 and α -chlorocarbonyl compounds (chloroacetone and 2-chloroacetophenone) in MeOH under neutral conditions, we assume that the HCl generated from the first step in Scheme 1, acted as an acid catalyst for the [5,5]-shifts of 3a and 3c to 5a and 5c, respectively.

On the other hand, we examined the condensation of 2,5-dithiobiurea (2) with α -chlorocarbonyl compounds according to the same way as shown in Scheme 1. We have also found the condensation of 2^{2b-d} with α -chlorocarbonyl compounds at 30–55°C, followed by rearrangement of the resultant N,N'-bis(2-thiazolyl)hydrazine derivatives (4) under the acidic condition to give 5,5'-bis(2-aminothiazole) derivatives (6), as illustrated in Scheme 2. After the rearrangement of 4 to 6, disappearance of thiazole 5-H signal in the region δ 6.2–7.0 of 4 is characteristic. The overall result of pathway (A) in Scheme 2 was consistent with that of pathway (B) and is summarized in Table 2.

In another interesting Hantzsch synthesis, the reaction of 2,5-dibromo-3,4-hexanedione or 1,4-dibromo-2,3-butanedione with 2 equiv. of thiourea in refluxing MeOH (Eq. (1)) gave the corresponding 4,4'-bis(2-aminothiazole) derivative (**7a** or **7b**). Compound **7a** or **7b** was in contrast to **6d** or **6e** in Scheme 2 (Eq. (2)) in terms of site selectivity, respectively.

Scheme 2.

 $\label{eq:table 2} Table\ 2$ The reactions of 2,5-dithiobiureas (2) with $\alpha\text{-chlorocarbonyl compounds}$

2	chlorocarbonyla	reaction condition	product	yield,%b	1 H-nmr data (DMSO- d_{6}), δ
2a	0=	MeOH/40 ℃/12 h	4a	77	(CDCl ₃); 2.25 (s, 6H), 3.25 (s, 6H), 6.2 (s, 2H)
	Me	H ₂ O/120 ℃/14 h	6a	81 (80)°	2.0 (s, 6H), 2.77 (d, 6H), 7.45 (q, 2H)
	OMe	H ⁺ /H ₂ O/45 ℃/15 h	4b	70	(CDCl ₃); 3.3 (s, 6H), 6.7 (d, 2H), 7.25 (d, 2H)
	MeO	H⁺/H ₂ O/125℃/17 h	6b	78 (77)°	2.85 (d, 6H), 6.95 (s, 2H), 7.6 (q, 2H)
	0	MeOH/30 ℃/12 h	4c	75	(CDCl ₃); 3.4 (s, 6H), 6.9 (s, 2H), 7.3-7.9 (m, 10H)
	Ph	MeOH/75 ℃/13 h	6c	88 (85)°	2.93 (s, 6H), 3.4 (bs, 2H), 7.2-7.7 (m, 10H)
2b	0=	MeOH/40 ℃/12 h	4d	74	2.1 (s, 6H), 5.75 (s, 2H), 6.25 (s, 2H)
	Me CI	H₂O/120 ℃/14 h	<u>6d</u>	75 (75)°	1.95 (s, 6H), 6.8 (s, 4H)
	OMe I	H ⁺ /H ₂ O/55 ℃/15 h	4e	70	6.85 (d, 2H), 7.25 (d, 2H), 7.7 (bs, 2H)
	MeO CI	H ⁺ /H ₂ O/130 ℃/16 h	<u>6e</u>	58 (52)°	7.0 (s, 2H), 7.6 (bs, 4H)
	0	MeOH/30 ℃/12 h	4f	78	7.0 (s, 2H), 7.3-7.8 (m, 12H)
	Ph	MeOH/75 ℃/13 h	6f	78 (77)°	6.8 (bs, 4H), 7.2-7.7 (m, 10H)

 $^{^{}a}2$ equivalents of α -chlorocarbonyl compounds were used. b yield isolated after column chromatography.

cyield rearranged to $\bf 6$ by refluxing $\bf 4$ (H^+/H_2O).

Those reactions are simple and effective, and also possess interesting potential as a new method in polythiazole synthesis.

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- 2. (a) The procedure for the synthesis of *N*-anilinothiourea (1) is as follows: To 20 ml of H₂O was dissolved 5 mmol of phenylhydrazine hydrochloride and 5 mmol of NH₄SCN. The mixture was refluxed for 12 h at 120°C and the solvent was evaporated under reduced pressure. Pale brown solid was recrystallized from H₂O. Yield: 88%, ¹H NMR (DMSO-*d*₆) δ: 6.7–7.2 (m, Ph), 7.5 (bs, NH), 9.25 (s, NH), ¹³C NMR (DMSO-*d*₆) δ: 113, 118.7, 128, 147.5, 182. (b) Typical procedure for the synthesis of 3,4-dimethyl-2,5-dithiobiurea (2a) is as follows: To 20 ml of H₂O was dissolved 10 mmol (1.33 g) of 1,2-dimethylhydrazine dihydrochloride and 20 mmol (1.52 g) of NH₄SCN. After the mixture was refluxed for 15 h at 120°C, solid was filtered and washed 3 times with 20 ml of H₂O and dried. Yield: 75%, mp=208–210°C (dec.), ¹H NMR (DMSO-*d*₆) δ: 3.3 (s, 6H) 7.7 (bs, NH₂), ¹³C NMR (DMSO-*d*₆) δ: 36.5, 181.5 (c) For 2,5-dithiobiurea (2b), yield: 58%, mp=209–211°C (lit.^{2d} 212°C), ¹H NMR (DMSO-*d*₆) δ: 7.6 (bs, NH₂), 9.4 (s, NH), ¹³C NMR (DMSO-*d*₆) δ: 180. (d) Compound 2b is also commercially available from Aldrich Chem. Co.
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- 4. NMR data for **7a** and **7b**. (a) **7a**; yield: 85%, mp=263–265°C, 1 H NMR (DMSO- d_{6}) δ : 2.35 (s, 6H), 6.60 (s, NH₂), 13 C NMR (DMSO- d_{6}) δ : 13.5, 117.5, 141, 163.5. (b) **7b**; yield: 88%, mp=246–248°C, 1 H NMR (DMSO- d_{6}) δ : 6.62 (s, 2H), 6.95 (s, NH₂), 13 C NMR (DMSO- d_{6}) δ : 102.5, 146.5, 168.